

STERESELECTIVE SYNTHESIS OF METHOXY SUBSTITUTED
1,2,3,4,4a,5,10,10a-OCTAHYDROBENZO[g]QUINOLINES

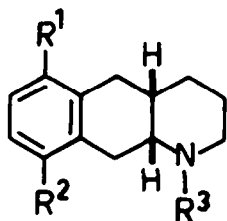
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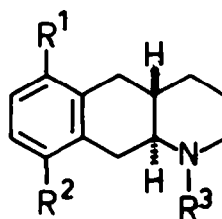
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Abstract. - The preparation of the *cis*- and *trans*-isomers of 6- and 9-methoxy-1,2,3,4,4a,5,10,10a-octahydrobenzo[g]-quinoline is reported. The syntheses involved reductions of cyclic iminium chlorides, which afforded the diastereomers conveniently and in good yields. Small *cis/trans* ratios were obtained with NaCNBH_3 as the reducing agent. Catalytic hydrogenation using PtO_2 in THF or *t*-BuOH gave the largest *cis/trans* ratios. The *N*²-benzyl derivatives were prepared to permit determination of relative stereochemistries.

Derivatives of 1,2,3,4,4a,5,10,10a-octahydrobenzo[g]quinoline (OHB[g]Q) may be considered as rigid phenethylamine congeners. Such compounds frequently possess impressive biological properties and oxygenated OHB[g]Q derivatives have demonstrated dopaminergic^{1,2} or analgetic and narcotic activities.³ Previously, oxygenated OHB[g]Q derivatives have been prepared by i) cyclization of 2-(methoxybenzyl)piperidine-3-carboxylic acids,^{1,2,4} ii) cyanoethylation of a 3-methoxycarbonyl-2-tetralone followed by reductive cyclization,⁵ and iii) cyclization of 2-(methoxybenzyl)- α,α -dimethyl-3-piperidinemethanol.³ In general, these reactions proceed in poor yield and/or are limited to certain substitution patterns. The present report describes a new, convenient and stereoselective synthesis of the *cis*- and *trans*-isomers of 6- and 9-methoxy-OHB[g]Q (1 and 2, respectively).



- 1a: $\text{R}^1 = \text{OCH}_3$, $\text{R}^2 = \text{R}^3 = \text{H}$
1b: $\text{R}^2 = \text{OCH}_3$, $\text{R}^1 = \text{R}^3 = \text{H}$
9a: $\text{R}^1 = \text{OCH}_3$, $\text{R}^2 = \text{H}$, $\text{R}^3 = \text{CH}_2\text{Ph}$
9b: $\text{R}^1 = \text{H}$, $\text{R}^2 = \text{OCH}_3$, $\text{R}^3 = \text{CH}_2\text{Ph}$



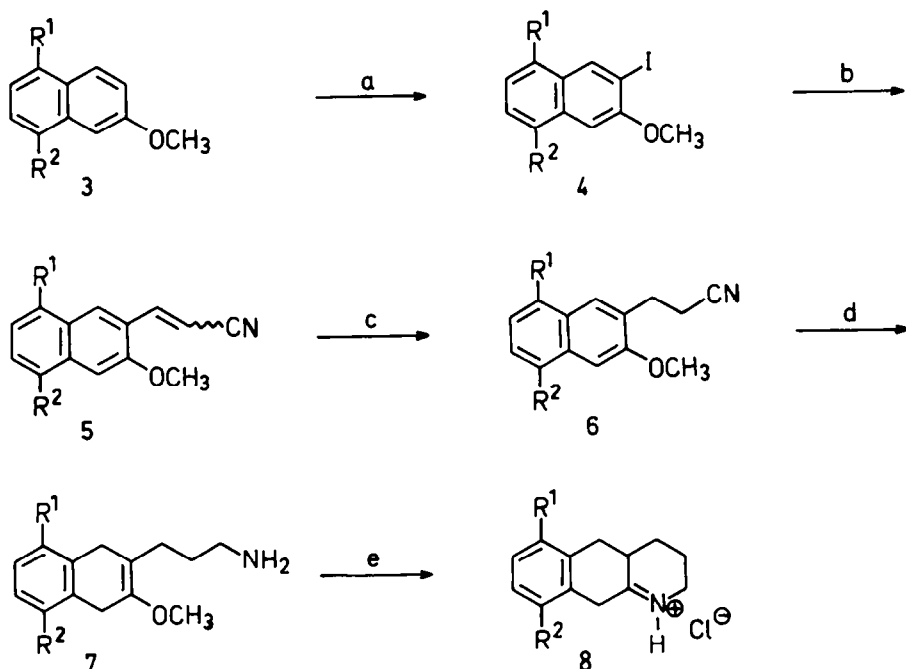
- 2a: $\text{R}^1 = \text{OCH}_3$, $\text{R}^2 = \text{R}^3 = \text{H}$
2b: $\text{R}^2 = \text{OCH}_3$, $\text{R}^1 = \text{R}^3 = \text{H}$
10a: $\text{R}^1 = \text{OCH}_3$, $\text{R}^2 = \text{H}$, $\text{R}^3 = \text{CH}_2\text{Ph}$
10b: $\text{R}^1 = \text{H}$, $\text{R}^2 = \text{OCH}_3$, $\text{R}^3 = \text{CH}_2\text{Ph}$

RESULTS AND DISCUSSION

Synthesis of iminium chlorides 8a and 8b. The synthetic sequence giving 8a and 8b is depicted below.

We have recently reported that ortho-lithiation of dimethoxynaphthalene 3a occurs regioselectively at C7 and that 3b regioselectively gives ortho-lithiation in the C6-position.⁶ Thus, lithium anions of 3a and 3b were treated with iodine to produce iodides 4a and 4b. As expected, 4b was isolated in high yield after iodination of 3b. Compound 4a, however, was obtained in only 57% yield from 3a. In methylations of ortho-lithiated 3a we observed the formation of an 8:2 mixture of 7- and 5-methylated isomers.⁶ It is therefore noteworthy that attempts to detect or isolate the 5-iodo- or any other isomer of 4a failed.

Heck vinylation⁷ of the iodinated isomers 4a and 4b, using acrylonitrile, triethylamine, and a catalytic amount of Pd(OAc)₂ in MeCN gave 4:1 mixtures (¹H NMR) of E- and Z-isomers 5 in excellent yields. The stereochemical assignments of the E- and Z-isomers were mainly based on the magnitudes of the vinylic proton coupling constants (E: $J=16.7, 16.9$ Hz; Z: $J=12.2$ Hz). The E-isomers could be isolated by recrystallization. However, in preparative runs, the crude E/Z mixtures were converted directly to the saturated nitriles 6a and 6b (H₂, 10% Pd/C, MeOH/CH₂Cl₂). Thus, pure 6a and 6b were obtained in > 92 % yield from 4a and 4b, respectively, after recrystallization.



a: R¹ = OCH₃, R² = H

b: R¹ = H, R² = OCH₃

Reagents: a) BuLi/THF, I₂; b) CH₂=CHCN, N(ET)₃, Pd(OAc)₂, MeCN; c) H₂, Pd/C, MeOH/CH₂Cl₂; d) Na, EtOH; e) HCl(aq)/MeOH.

In the next step, both the more substituted ring and the cyano group were reduced (Na, EtOH) to give 7a and 7b as the sole products. Cyano groups do not appear to be compatible with ordinary Birch conditions⁸ but no reductive cleavage of the cyano group was observed under the present conditions. Solvolysis of enol ethers 7a and 7b furnished iminium chlorides 8a and 8b. We were never able to observe the intermediate ketones. Thus, the formation of the piperidine ring and the subsequent dehydration seem to occur spontaneously.

Reductions of 8a and 8b. The iminium chlorides 8a and 8b served as synthetic intermediates both for the cis- and trans-OHB[g]Q derivatives. NaCNBH₃ reduction of 8a or 8b in MeOH at pH 6 produced predominantly the trans-isomers 2a and 2b (cis/trans ratios = 1:9). The pure trans isomers were obtained by fractional crystallization of the hydrochlorides.

The stereochemical outcome of the catalytic hydrogenation of 8a and 8b varied with catalyst and solvent (Table I). In addition, hydrogenation of 8a consistently gave larger cis/trans ratios than 8b. The cis/trans product ratios shown in Table I increase with increasing dielectric constants of the aprotic solvents (EtOAc, THF) but decrease with increasing dielectric constants of the neutral protic solvents (*t*-BuOH, MeOH). Thus, the present results agree with previously observed stereochemistries of carbonyl hydrogenations.⁹ Augustine *et al.*¹⁰ have demonstrated that most of these effects can be rationalized if both steric and electronic solvent effects are considered; the ability of the solvent to compete for catalyst sites considerably affects the nature of the catalyst and thus the rate as well as the selectivity of the reduction.

The largest cis/trans ratios were obtained with PtO₂ in *t*-BuOH or in THF (Table I). Under these conditions, pure 1a·HCl was obtained in 51 % yield from 8a after recrystallization. However, pure 1b could not be conveniently prepared from 8b by this procedure. Instead, the mixture of 1b and 2b was *N*-benzylated and the *N*-benzyl derivative 9b was then separated from 10b by chromatography on aluminum oxide followed by fractional crystallization. Isomerically pure 1b was finally obtained, in 31 % overall yield from 8b, by hydrogenolysis of 9b.

Table I. Product ratios in catalytic hydrogenations of iminium chlorides 8a and 8b.

Catalyst	Solvent	Isomeric ratio ^a	
		1a : 2a	1b : 2b
Pd(C) 10%	MeOH	29 : 71	10 : 90
Pd(C) 10%	<i>t</i> -BuOH	41 : 59	12 : 88
PtO ₂	HOAc	48 : 52	25 : 75
PtO ₂	MeOH	42 : 58	33 : 67
PtO ₂	EtOAc	62 : 38	37 : 63
PtO ₂	<i>t</i> -BuOH	75 : 25	48 : 52
PtO ₂	THF	81 : 19	47 : 53

^aBased on GC-analysis of trifluoroacetylated crude reaction products.

The consistent difference in stereoselectivity in reductions of 8a and 8b is probably not due to haptophilic effects¹¹ or to other effects related to heteroatom coordination.¹² Instead, the difference in stereoselectivities may be related to a greater tendency of 8b, as compared to 8a, to undergo imine/enamine isomerization; the styrenic enamine (enammonium ion) corresponding to 8b should be more stable than that formed from 8a due to conjugative stabilization between the 9-methoxy

group and the double bond. This interpretation is supported by the fact that Pd(C) gave smaller cis/trans ratios than PtO₂; Pd is known to isomerize double bonds to a larger extent than Pt.¹³

Structural assignments of 1 and 2. ¹H NMR spectral studies of the N-benzyl derivatives 9 and 10 made it possible to assign the stereochemistries of 1 and 2. In 90 MHz ¹H NMR spectra of CDCl₃ solutions of 10a and 10b, the α -hydrogens of the N-benzyl groups appeared as AB quartets centered at δ 3.74 and 3.77 ppm, respectively, with large chemical shift differences (0.92 and 1.11 ppm, respectively) between the A and B portions. In contrast, the corresponding hydrogens of 9a and 9b appeared as apparent singlets at δ 3.68 and 3.72 ppm, respectively. It has been demonstrated that the AB quartet due to the diastereotopic α -hydrogens of N-benzyl substituted trans-OHB[gl]Q derivatives and related compounds have a larger chemical shift difference between the A- and B- portions than the corresponding cis-derivatives.⁴ Thus, 1 and 9 were assigned the cis-stereochemistry and 2 and 10 should be trans-isomers.

NMR-Spectra of CD₃OD solutions of the hydrochlorides of 9 and 10 gave spectra similar to those of the corresponding bases in CDCl₃. However, in ¹H NMR spectroscopy, the benzylic protons of the hydrochlorides of 9a and 9b appeared as two singlets in a 7:1 ratio and in the proton decoupled ¹³C NMR spectra of these compounds several carbons appeared as two signals exhibiting approximately the same difference in intensity (7:1), indicating the presence of two conformations in a slow (NMR-time scale) equilibrium.

Concluding remarks. The syntheses presented here are convenient to perform in multi gram-scale and the overall yields of the OHB[gl]Q derivatives 1a, 1b, 2a and 2b are quite acceptable (22 %, 22 %, 27 %, and 25 %, respectively, as calculated from the easily available dimethoxynaphthalene derivatives 3a and 3b). A number of compounds of potential biological interest can be prepared from these new OHB[gl]Q derivatives.

EXPERIMENTAL SECTION

General Comments. Dry THF was distilled from Na metal in the presence of benzophenone under dry N₂. Other chemicals were used as received. Melting points (uncorrected) were determined in open glass capillaries on a Thomas-Hoover apparatus. ¹H and ¹³C NMR spectra were recorded on a JEOL FX 90Q spectrometer at 20 °C and were referenced to internal Me₄Si. ¹³C NMR data of the hydrochlorides of compound 1, 2, 8, 9 and 10 in CD₃OD are given in Table II. IR spectra were recorded on a Perkin-Elmer 157G spectrophotometer. Mass spectra were recorded at 70 eV on a LKB 9000 spectrometer using a direct insertion probe. GC was performed on a Varian 2700 instrument with a flame ionization detector. Thin-layer chromatography (TLC) was carried out on aluminium sheets precoated with silica gel 60 F₂₅₄ (0.2mm), or for amines, on Al₂O₃ 60 F₂₄₅ neutral (Typ E), E. Merck. Elemental analyses were performed by Mikro Kemi AB, Uppsala, Sweden.

Iodination of 3b. Preparation of 3,5-Dimethoxy-2-iodonaphthalene (4b). A solution of butyllithium in hexane (1.5 M; 177 mL, 0.27 mol) was added to a stirred solution of 3b (50 g, 0.27 mol) in dry THF (500 mL) kept under N₂ at -78 °C. After stirring at rt over night, the reaction mixture was cooled to -78 °C and a solution of I₂ (71 g, 0.27 mol) in dry THF (100 mL) was added dropwise. When the addition was complete, a saturated aqueous solution of NH₄Cl (200 mL) was added followed by a saturated aqueous solution of Na₂S₂O₃ (100 mL). The volatiles were evaporated and ether was added to the residue. The ether layer was washed with water, dried (MgSO₄) and concentrated to afford 76.9 g (92 %) of 4b which was pure according to NMR and GC. An analytical sample was prepared by recrystallisation from ether; mp 125-126 °C; R_f 0.77 (ether/light petroleum 1:1); ¹H NMR (CDCl₃) δ 8.26 (s, C4-H), 7.46 (s, C1-H), 7.28 - 7.18 (m, C6-H, C8-H), 6.90 - 6.70 (m, C7-H), 2.98 (s, OMe's); ¹³C NMR (CDCl₃) δ 154.85, 154.30 (C3, C5), 138.76 (C1), 131.19 (C8a), 126.19 (C4a), 124.25 (C7), 118.84 (C8), 104.60 (C6), 100.19 (C4), 88.76 (C2), 56.36, 55.44 (OMe's); mass spectrum, m/z (relative intensity) 314 (100, M⁺), 299 (42, M⁺-CH₃), 127 (19, I⁺). Anal. Calc for C₁₂H₁₁O₂I: C, 45.9 H, 3.5. Found: C, 45.9 H, 3.5.

Iodination of 3a. Preparation of 2,5-Dimethoxy-3-iodonaphthalene (4a). Compound 4a was prepared from 3a (50 g, 0.27 mol) by the above procedure. Ether was added to the concentrated residue and pure 4a (47.6 g, 57%) crystallized on standing: mp 110.5–112 °C; *R_f* 0.74 (ether/light petroleum 1:1); ¹H NMR (CDCl₃) δ 8.72 (s, C4-H), 7.45–7.28 (m, C6-H, C8-H), 7.01 (s, C1-H), 6.72–6.54 (m, C7-H), 3.95 (s, OMe's). ¹³C NMR (CDCl₃) δ 155.47, 154.36 (C2, C5), 135.43 (C8a), 122.05 (C4a), 133.85 (C4), 127.18 (C7), 118.87 (C8), 105.15 (C6), 102.41 (C1), 86.75 (C3), 56.21, 55.37 (OMe's); mass spectrum *m/z* (relative intensity) 314 (100, M⁺), 299 (14, M⁺-CH₃), 127 (21, I⁻). Anal. Calcd for C₁₂H₁₁O₂I: C, 45.9 H, 3.5. Found: C, 45.8 H, 3.5.

Heck vinylation of 4b. Preparation of (E)- and (Z)-2-(2-cyanoethyl)-3,5-dimethoxynaphthalene (5b). A mixture of 4b (30 g, 95.5 mmol), acrylonitrile (7.6 g, 143.2 mmol), Pd(OAc)₂ (214 mg, 0.95 mmol), triethylamine (9.7 g, 95.5 mmol) and MeCN (40 mL) was heated in a capped Pyrex flask at 120 °C for 3 d. The cooled reaction mixture was filtered (Celite), CH₂Cl₂ was added and the solution was washed with 1 M HCl and with brine, dried (MgSO₄), filtered and concentrated. The residue was dissolved in a small amount of CH₂Cl₂ and eluted through a short silica column with ether/light petroleum 1:1 as eluant yielding 21.9 g (96%) of a 4:1 mixture of (E)- and (Z)-5b. An analytical sample of (E)-11b was obtained by recrystallization twice from MeOH; (E)-5b; mp 139.5–140.5 °C; *R_f* 0.58 (ether/light petroleum 1:1); IR: C≡N stretch √ 2210; ¹H NMR (CDCl₃) δ 7.65 (s, C4-H), 7.60 (d, J = 16.7 Hz, Cα-H), 7.48 (s, C1-H), 7.35–6.70 (m, C6-H, C7-H, C8-H), 6.15 (d, J = 16.7 Hz, Cβ-H), 3.97, 3.93 (OMe's). ((Z)-11b: δ 5.52 (d, J = 12.2 Hz, Cβ-H)); ¹³C NMR (CDCl₃) δ 155.25, 154.11 (C3, C5), 146.76 (Cβ), 129.62 (C1), 128.91, 124.09 (C4a, C8a), 127.24 (C2), 124.43 (C7), 120.42 (C8), 118.90 (CN), 105.72 (C6), 100.74 (C4), 98.12 (Cα), 55.50 (OMe's); mass spectrum *m/z* (relative intensity) 239 (100, M⁺), 224 (61, M⁺-CH₃). Anal. Calcd for C₁₅H₁₃NO₂: C, 75.3 H, 5.5 N, 5.9. Found: C, 75.2 H, 5.5 N, 5.9.

Heck vinylation of 4a. Preparation of (E)- and (Z)-3-(2-cyanoethyl)-2,5-dimethoxynaphthalene (5a). This mixture was prepared from 4a (30 g, 95.5 mmol) by the above procedure. Yield 20.6 g (90%); (E)-5a; mp 98–99.5 °C; *R_f* 0.53 (ether/light petroleum 1:1); IR: C≡N stretch √ 2210; ¹H NMR (CDCl₃) δ 8.24 (s, C4-H), 7.68 (d, J = 16.8 Hz, Cα-H), 7.56–7.15 (m, 2H), 7.04 (s, C1-H), 6.72–6.60 (m, 1H), 6.20 (d, J = 16.9 Hz, Cβ-H), 3.92, 3.85 (s's, OMe's); ((Z)-5a: δ 5.50 (d, J = 12.2 Hz, Cα-H)); ¹³C NMR (CDCl₃) δ 155.96 (C2, C5), 146.91 (Cβ), 136.69 (C8a), 122.89 (C4a), 128.63 (C4), 124.52 (C7), 120.80 (C3), 118.93 (CN), 118.78 (C8), 105.68 (C6), 102.57 (C1), 97.41 (Cα), 55.41 (OMe's); mass spectrum *m/z* (relative intensity) 239 (100, M⁺), 196 (39). Anal. Calcd for C₁₅H₁₃NO₂: C, 75.3 H, 5.5 N, 5.9. Found: C, 75.4 H, 5.6 N, 5.8.

2-(2-Cyanoethyl)-3,5-dimethoxynaphthalene (6b). A crude mixture of (E)- and (Z)-5b (5.0 g, 20.9 mmol) was dissolved in CH₂Cl₂ (20 mL) and MeOH (80 mL) and the mixture was hydrogenated over 10% Pd(C) at 3 atm. The reaction mixture was filtered (Celite) and concentrated. Recrystallization of the residue from MeOH afforded 4.7 g (93 %) of pure 6b; mp 79–80 °C; *R_f* 0.53 (ether/light petroleum 1:1); IR: C≡N stretch √ 2240; ¹H NMR (CDCl₃) δ 7.57, 7.52 (s's, C1-H, C4-H), 7.41–7.10 (m, C6-H, C8-H), 6.85–6.72 (m, C7-H), 3.99, 3.97 (s's, OMe's), 3.20–2.95 (m, 2H), 2.30–2.10 (m, 2H); ¹³C NMR (CDCl₃) δ 155.41, 154.26 (C3, C5), 129.50, (C8a), 128.75 (C1), 128.20 (C2), 125.73 (C4a), 123.78 (C7), 110.67 (C8), 119.58 (CN), 104.17 (C6), 99.88 (C4), 55.37, 55.31 (OMe's), 27.42 (Cβ), 17.45 (Cα); mass spectrum *m/z* (relative intensity) 241 (100, M⁺), 226 (39, M⁺-CH₃), 201 (15, M⁺-CH₂CN). Anal. Calcd for C₁₅H₁₅NO₂: C, 74.7 H, 6.3 N, 5.8. Found: C, 74.8 H, 6.1 N, 5.8.

3-(2-Cyanoethyl)-2,5-dimethoxynaphthalene (6a). Compound 6a was prepared from 5a (5.0 g, 20.9 mmol) by the above procedure. Yield 4.9 g (97%); mp 80.5–82 °C; *R_f* 0.51 (ether-light petroleum 1:1); IR: C≡N stretch √ 2240; ¹H NMR (CDCl₃) δ 8.02 (s, C4-H), 7.40–7.22 (m, 2H), 7.03 (s, C1-H), 6.71–6.57 (m, 1H), 3.93, 3.88 (s's, OMe's), 3.20–2.95 (m, 2H), 2.80–2.14 (m, 2H); ¹³C NMR (CDCl₃) δ 156.18, 155.28 (C2, C5), 135.24 (C8a), 126.93 (C4a), 126.56 (C4), 123.41 (C7), 120.29 (C3), 119.61 (CN), 118.84 (C8), 105.10 (C6), 102.26 (C1), 55.37, 55.22 (OMe's), 27.67 (Cβ), 17.57 (Cα); mass spectrum *m/z* (relative intensity) 241 (89, M⁺), 201 (100, M⁺-CH₂CN). Anal. Calcd for C₁₅H₁₅NO₂: C, 74.7 H, 6.3 N, 5.8. Found: C, 74.6 H, 6.4 N, 5.8.

Sodium reduction of 6b. Preparation of 2-(3-aminopropyl)-3,5-dimethoxy-1,4-dihydro-naphthalene (7b). Slices of Na (40 g, 1.74 mol) were added during ten minutes to a boiling solution of 6b (10.0 g, 41.4 mmol) in EtOH (400 mL) under N₂. After one h the heating was interrupted and water (100 mL) and NH₄Cl (95 g) were added carefully. The mixture was filtered and the volatiles were evaporated. The residue was partitioned between water and ether. The ether layer was washed several times with water, dried (K₂CO₃) and concentrated to afford 8.8 g (86%) of almost pure 7b; ¹H NMR (CDCl₃) δ 7.25–6.58 (m, 3H), 3.82, 3.62 (s's, OMe's), 3.40 (app s, 4H), 2.80–1.20 (m, 3H); ¹³C NMR (CDCl₃) δ 156.77 (C5), 145.74 (C3), 135.21 (C8a), 122.64 (C4a), 126.47 (C7), 120.01 (C8), 113.74 (C2), 106.86 (C6), 56.24, 55.19 (OMe's), 41.66 (Cα), 33.42, 31.46, 25.76, 24.09 (C1, C4, Cβ, Cγ); mass spectrum *m/z* (relative intensity) 247 (16, M⁺), 232 (56, M⁺-CH₃), 215 (74).

Sodium reduction of 6a. Preparation of 3-(3-Aminopropyl)-2,5-dimethoxy-1,4-dihydro-naphthalene (7a). Compound 7a was prepared from 6a (15.0 g, 62.2 mmol) by the above procedure. Yield 14.3 g (93%); $^1\text{H NMR}$ (CDCl_3) δ 7.25-6.97 (m, 1H), 6.85-6.52 (m, 2H), 3.81, 3.59 (s's, OMe's), 3.70-3.18 (m, 4H), 2.80-2.12 (m, 4H), 1.80-1.20 (m, 4H); $^{13}\text{C NMR}$ (CDCl_3) δ 156.77 (C5), 145.15 (C2), 134.96, (C4a), 122.98 (C8a), 126.50 (C7), 120.38³ (C8), 114.39 (C3), 106.98 (C6), 56.27, 55.19 (OMe's), 41.57 (Ca), 31.41, 29.19, 28.17, 25.94, (C1, C4, C8, Cg); mass spectrum m/z (relative intensity) 247 (12, M⁺), 232 (39, M-CH₃), 215 (63).

Solvolysis of 7b. Preparation of 9-Methoxy-2,3,4,4a,5,10-hexahydrobenzo[*g*]quinolinium chloride (8b). A solution of 7b (20 g, 80.9 mmol) in MeOH (400 mL) and 12 M HCl (70 mL) was heated to reflux under N₂. After four h, the solution was concentrated to dryness. The residue was recrystallized from EtOH-ether yielding 18.1 g (83%) of pure 8b; mp 217-219 °C; $^1\text{H NMR}$ (MeOH-d_4) δ 7.35-7.12 (m, 1H), 6.95-6.75 (m, 2H), 3.84 (s, OMe), 3.85-3.65 (m, 2H), 3.35-1.50 (m, 9H); mass spectrum m/z (relative intensity), 215 (100, M⁺), 200 (72, M⁺-CH₃). Anal. calcd for C₁₄H₁₇NO·HCl: C, 66.8 H, 7.2 N, 5.6. Found: C, 66.7 H, 7.0 N, 5.4.

Solvolysis of 7a. Preparation of 6-Methoxy-2,3,4,4a,5,10-hexahydrobenzo[*g*]quinolinium chloride (8a). Compound 8a was prepared from 7a (6.3 g, 25.5 mmol) by the same procedure but instead of being recrystallized the hydroscopic crude 8a was triturated with ether. Yield 6.0 g (94%); mp 189-193 °C; $^1\text{H NMR}$ (MeOH-d_4) δ 7.31-7.13 (m, 1H), 6.90-6.70 (m, 2H), 3.82 (s, OMe), 3.85-1.30 (m, 11H); mass spectrum m/z (relative intensity) 215 (100, M⁺), 187 (13), 186(18). Anal. Calcd for C₁₄H₁₇NO·HCl·1/2H₂O: C, 64.5 H, 7.3 N, 5.4. Found: C, 64.6 H, 7.7 N, 5.0.

Sodium cyanoborohydride reduction of the iminium chlorides. Preparation of trans-9-methoxy-1,2,3,4,4a,5,10,10a-octahydrobenzo[*g*]quinolinium chloride (2b·HCl). NaCNBH₃ (0.43 g, 7.8 mmol) was added in portions to a solution of 8b (3.0 g, 11.9 mmol) in MeOH (50 mL) coloured by a small amount of methyl red. The red colour was maintained during the reaction by addition of methanolic HCl when necessary. After two h the mixture was acidified to pH 2 and concentrated. The residue was partitioned between 1M NaOH and ether. The ether layer was dried (K₂CO₃) filtered and concentrated. Etheral HCl was added to an etheral solution of the amine and the resulting hydrochloride salt was recrystallized twice from EtOH-ether affording 1.30 g (43%) of pure 2b·HCl; mp 265.5-266.5 °C; $^1\text{H NMR}$ (MeOH-d_4) δ 7.22-7.02 (m, 1H), 6.82-6.62 (m, 2H), 3.81 (s, OMe), 3.60-1.10 (m, 12H); mass spectrum m/z (relative intensity) 217 (89, M⁺), 134 (59), 82 (100). Anal. Calcd for C₁₄H₁₉NO·HCl: C, 66.3 H, 7.9 N, 5.5. Found: C, 66.0 H, 8.1 N, 5.6.

Preparation of trans-6-methoxy-1,2,3,4,4a,5,10,10a-octahydrobenzo[*g*]quinoline hydrochloride (2a·HCl). Compound 2a·HCl was prepared from 8a (10.0 g, 39.7 mmol) by the above procedure. Yield 6.12 g (61%); mp 276.5-278.5 °C; $^1\text{H NMR}$ (MeOH-d_4) δ 7.22-7.00 (m, 1H), 6.88-6.62 (m, 2H), 3.79 (s, OMe), 3.60-2.80 (m, 5H), 2.40-1.20 (m, 7H); mass spectrum m/z (relative intensity) 217 (66, M⁺), 134 (37), 82 (100). Anal. Calcd for C₁₄H₁₉NO·HCl: C, 66.3 H, 7.9 N, 5.5. Found: C, 66.4 H, 7.9 N, 5.5.

Catalytic hydrogenation of the iminium chlorides. Preparation of cis- and trans-1-benzyl-9-methoxy-1,2,3,4,4a,5,10,10a-octahydrobenzo[*g*]quinoline (1b and 2b). A suspension of 8b (10.0 g, 39.7 mmol) in t-BuOH (400 mL) was hydrogenated over PtO₂ at atmospheric pressure. When the hydrogen uptake had ceased, the catalyst was filtered off (Celite) and the volatiles were evaporated. The resulting 1:1 mixture of the two diastereomers 1b and 2b was benzylated prior to separation: The crude residue was dissolved in MeCN (200 mL), K₂CO₃ (16.5 g, 130 mmol) and benzyl chloride (7.5 g, 59.2 mmol) was added, and the mixture was stirred at rt over night. After filtration (Celite) and concentration of the reaction mixture, the mixture of 9b and 10b was separated by chromatography on an alumina column using ether-light petroleum 1:9 as eluant. Impure fractions containing excess of the 10b were converted to the hydrochloride salt and purified by recrystallization from EtOH. The free amine from the mother liquor containing mainly the cis-isomer was purified by column chromatography. The combined fractions of the trans-isomer were treated with etheral HCl. The resulting hydrochloride was recrystallized from EtOH-ether yielding 5.2 g (38%) of 10b·HCl; mp 216-218 °C; R_f 0.30 (ether-light petroleum 1:9); $^1\text{H NMR}$ (MeOH-d_4) δ 7.64-7.40 (m, 5H), 7.38-7.05 (m, 1H), 6.87-6.14 (m, 2H), 4.91, 4.14 (d's, J=13.2, Bz-CH₂) 3.86 (s, OMe), 3.72-1.10 (m, 12H); mass spectrum m/z (relative intensity) 307 (86, M⁺), 173 (50), 172 (47). Anal. Calcd for C₂₁H₂₅NO·HCl·1/2H₂O: C, 71.5 H, 7.9 N, 4.0. Found: C, 71.4 H, 7.9 N, 4.0. Compound 9b·HCl could not be recrystallized and was handled as the base. Recrystallization of 9b from ether-light petroleum yielded 5.1 g (42%) of pure 9b; mp 100-101 °C; R_f 0.47 (ether-light petroleum 1:9). A sample of the amorphous hydrochloride salt was prepared for spectral data; $^1\text{H NMR}$ (MeOH-d_4) δ 7.80-7.38 (m, 5H), 7.22-7.00 (m, 1H), 6.85-6.58 (m, 2H), 4.41 (s, Bz-CH₂), 3.87 (s, OMe), 3.40-1.25 (m, 12H). Additional peaks corresponding to the minor conformation; δ 4.65 (s, Bz-CH₂), 3.79 (s, OMe); mass spectrum m/z (relative intensity) 307 (71, M⁺), 173 (50), 172 (46), 91 (100). Anal. calcd for C₂₁H₂₅NO: C, 82.0 H, 8.2 N, 4.6. Found: C, 82.4 H, 8.4 N, 4.6.

Benzylation of 9b. Preparation of cis-9-methoxy-1,2,3,4,4a,5,10,10a-octahydrobenzo[*g*]quinolinium chloride (1b·HCl). Etheral HCl was added to an etheral solution of 9b (0.22 g, 0.72 mmol) and the precipitate was collected and dissolved in MeOH

Table II. ^{13}C Nuclear Magnetic Resonance Spectra of some 1,2,3,4,4a,5,10,10a-octahydrobenzo[*g*]quinolines and Related Derivatives.^a

Carbon	8a	8b	1a	1b	2a	2b	9a	9b	10a	10b
C2	45.92	46.11	45.71	45.55	44.38	43.64	53.18	53.15	47.07	46.85
C3	18.38 ^b	19.95	23.47	23.19	19.36	19.95	23.53	23.47	22.85 ^b	17.97
C4	19.80 ^b	24.40 ^b	30.51 ^b	28.54 ^b	24.27 ^b	25.48 ^b	30.57 ^b	27.73 ^b	23.75 ^b	23.16 ^b
C4a	29.34	38.39	36.16	35.73	30.85	31.35	37.06	37.12	29.37	33.69
C5	24.40	24.52 ^b	30.67 ^b	30.05 ^b	27.02 ^b	26.28 ^b	31.47 ^b	30.17 ^b	23.81 ^b	23.69 ^b
C5a	123.84	136.66	124.37	137.12	123.50	135.86	124.21	136.72	122.70	135.02
C6	157.66	121.03	158.34	121.56	158.80	122.27	158.22	121.22	158.71	122.02
C7	109.67	129.19	108.99	128.29	108.96	128.35	109.05	128.54	108.74	128.45
C8	129.13	109.54	128.14	108.43	128.15	108.43	128.26	108.59	128.01	108.40
C9	120.91	157.91	121.77	158.22	122.11	158.71	122.08	158.50	122.30	158.50
C9a	132.24	119.30	134.25	121.74	132.28	120.11	134.41	121.65	132.74	120.05
C10	<u>c</u>	35.45	34.31	36.29	31.81	30.36	33.01 ^b	37.12	33.45	35.15
C10a	193.61	193.86	57.97	58.19	54.36	53.98	66.46	66.86	59.36	59.57
OMe	55.93	56.05	55.78	55.78	55.78	55.78	55.87	55.87	55.63	55.81

^a The following peaks were assigned to the *N*-benzyl groups: 9a; 58.15 (Bz-CH₂), 130.70, 131.13 (C1',C4'), 132.68, 130.42 (C2',C3',C5',C6'), 9b; 57.86 (Bz-CH₂), 130.70, 131.13 (C1',C4'), 132.58, 130.42 (C2',C3',C5',C6'), 10a; 58.59 (Bz-CH₂), 130.67, 131.61 (C1',C4'), 132.24, 130.24 (C2',C3',C5',C6'), 10b; 58.52 (Bz-CH₂), 130.64, 130.98 (C1',C4'), 132.31, 130.21 (C2',C3',C5',C6'). ^b Ambiguous assignment. ^c Not observed.

(200 mL). Catalyst [Pd(C); 10%] was added and the mixture was hydrogenated at atmospheric pressure. The catalyst was filtered off (Celite) and the volatiles were evaporated. The residue was partitioned between ether and 1 M NaOH. The ether layer was dried (K_2CO_3), filtered and concentrated. The secondary amine thus obtained was converted to the hydrochloride and recrystallized from MeOH-ether to give 0.16 g (88%) of pure **1b**·HCl; mp 226.5-228 °C; 1H NMR (MeOH-*d*) δ 7.24-7.05 (m, 1H), 6.85-6.68 (m, 2H), 3.82 (s, OMe), 3.90-3.68 (m, C10a), 3.47-2.80 (m, 6H), 2.50-2.18 (m, C4a), 2.00-1.57 (m, 4H); mass spectrum m/z (relative intensity) 217 (99, M⁺), 134 (67), 82 (100). Anal. Calcd for $C_{14}H_{19}NO\cdot HCl$: C, 66.3 H, 7.9 N, 5.5. Found: C, 66.3 H, 8.2 N, 5.5.

Preparation of cis-6-methoxy-1,2,3,4,4a,5,10,10a-octahydrobenzo[*g*]quinoline hydrochloride (**1a**·HCl). Compound **1a**·HCl was prepared from **8a** (2.0 g, 7.94 mmol) by hydrogenation in *t*-BuOH (80 mL) with PtO_2 as catalyst using the same procedure as described above. Recrystallization of the crude salt from EtOH-ether afforded 1.0 g (51%) of pure **1a**·HCl; mp 218-220 °C; 1H NMR (MeOH-*d*) δ 7.24-7.06 (m, 1H), 6.83-6.70 (m, 2H), 3.82 (s, OMe), 3.80-3.62 (m, 1H), 3.35-3.00 (m, 4H), 2.90-2.68 (m, 2H), 2.50-2.25 (m, 1H), 2.00-1.60 (m, 4H); mass spectrum m/z (relative intensity) 217 (61, M⁺), 134 (40), 104 (25), 82 (100). Anal. Calcd for $C_{14}H_{19}NO\cdot HCl$: C, 66.3 H, 7.9 N, 5.5. Found: C, 66.2 H, 8.1 N, 5.4.

Benzylation of **1a**. Preparation of cis-1-benzyl-6-methoxy-1,2,3,4,4a,5,10,10a-octahydrobenzo[*g*]quinoline (**9a**). A mixture of **1a** (6.0 g, 27.6 mmol), K_2CO_3 (14.5 g, 82.8 mmol), benzyl chloride (5.2 g, 41.4 mmol) and MeCN (80 mL) was stirred under N_2 for 8 h. The mixture was filtered and concentrated. Ether was added and the solution was washed with water, dried (K_2CO_3), filtered and concentrated. Recrystallization of the base from MeOH afforded 7.7 g (91%) of pure **9a**; mp 81.5-83 °C; Rf 0.50 (ether-light petroleum 1:9). A sample of the amorphous hydrochloride was prepared for spectroscopy; 1H NMR (MeOH-*d*) δ 7.83-7.40 (m, 5H), 7.30-7.05 (m, 1H), 6.95-6.70 (m, 2H), 4.41 (s, Bz-CH₂), 3.76 (s, OMe), 4.00-2.98 (m, 5H), 2.90-2.40 (m, 3H), 2.10-1.40 (m, 4H). Additional peaks corresponding to the minor conformation; δ 4.58 (s, Bz-CH₂), 3.80 (s, OMe); mass spectrum m/z (relative intensity) 307 (68, M⁺), 91 (100). Anal. Calcd for $C_{21}H_{25}NO$: C, 82.0 H, 8.2 N, 4.6. Found: C, 82.5 H, 8.0 N, 4.6.

Benylation of **2a**. Preparation of trans-1-benzyl-6-methoxy-1,2,3,4,4a,5,10,10a-octahydrobenzo[*g*]quinolinium chloride (**10a**·HCl). Compound **10a** was prepared from **2a** (7.5 g, 34.5 mmol) by the above procedure. The amine was converted to the hydrochloride and recrystallized from EtOH-ether yielding 11.2 g (94%) of pure **10a**·HCl; mp 235-237 °C; Rf 0.28 (ether-light petroleum 1:9); 1H NMR (MeOH-*d*) δ 7.68-7.40 (m, 5H), 7.38-7.05 (m, 1H), 6.90-6.70 (m, 2H), 4.89, 4.15 (d's, J=13.0 Hz, Bz-CH₂), 3.80 (OMe), 3.72-2.80 (m, 5H), 2.50-1.20 (m, 7H); mass spectrum m/z (relative intensity) 307 (71, M⁺), 91 (100). Anal. Calcd for $C_{21}H_{25}NO\cdot HCl$: C, 73.4 H, 7.6 N, 4.1. Found: C, 73.2 H, 7.3 N, 4.0.

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REFERENCES AND NOTES

1. Cannon, J.G.; Hamer, R.L.; Ilhan, M.; Bhatnagar, R.K.; Long, J.P. *J. Med. Chem.* **27**, 190 (1984); Cannon, J.G.; Amoo, V.E.D.; Long, J.P.; Bhatnagar, R.K.; Flynn, J.R. *ibid.* **29**, 2529 (1986).
2. R. Nordmann, R.; Petcher, T.J. *J. Med. Chem.* **28**, 367 (1985).
3. Michne, W.F.; Albertson, N.F. *J. Med. Chem.* **12**, 402 (1969), *ibid.* **13**, 522 (1970).
4. Walsh, D.A.; Smismann, E.E. *J. Org. Chem.* **39**, 3705 (1974).
5. Cannon, J.G.; Lee, T.; Beres, J.A.; Goldman, H.D. *J. Heterocycl. Chem.* **17** 1633 (1980).
6. Johansson, A.M.; Mellin, C.; Hacksell, U. *J. Org. Chem.* **51**, 5252 (1986).
7. Heck, R.F. *Org. React.* **27**, 345 (1982).
8. House, H.O. *Modern Synthetic Reactions*, 2:ed ed., W.A. Benjamin, Inc., Menlo Park, California, p.215 (1972).
9. For general discussions of effects of solvents and catalysts on the stereochemistry of hydrogenations, see Augustine, R.L. *Adv. Catal.* **25**, 63 (1976), and Rylander, P.N. In "Catalysis in Organic Synthesis"; Jones, W.H., Ed; Academic Press: New York, p 155 (1980).
10. Augustine, R.L.; Warner, R.W.; Melnick, M.J. *J. Org. Chem.* **49**, 4853 (1984).
11. For discussions of the haptophilic effect, see for example: Thomson, H.W.; Wong, J.K. *J. Org. Chem.* **50**, 4270 (1985), and Thompson, H.W.; Naipawer, R.E. *J. Am. Chem. Soc.* **95**, 6379 (1973).
12. Mundy, B.P.; Theodore, J.J. *J. Am. Chem. Soc.* **102**, 2005 (1980).
13. See for example Augustine, R.L.; Yaghmaie, F.; Van Peppen, J.F. *J. Org. Chem.* **49**, 1865 (1984).